

Selenium concentration in milks

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In this review article, papers on selenium concentrations in cow and human milk are reviewed in order to identify the main factors that affect these concentrations as well as the Se intake of lactating infants. Selenium intake and Se status of the mother seem to be the main factors that influence Se concentrations in human milk. The progression of lactation can diminish the Se concentrations in human milk. Most authors have reported Se levels in human milk greater than Se levels in available milk formulas for infants from the same country. Thus, higher Se intakes in the breast-fed infant than Se intakes in formula-fed infants have been observed. Also, the Se compounds in breast milk seem to be more biologically available for infant nutrition than those in formulas.

NOTATION

GSH-Px Glutathione peroxidase

INAA Instrumental neutron activation analysis
RNAA Radiochemical neutron activation analysis

EAAS Electrothermal atomic absorption

spectrometry

HG-AAS Hydride-generation atomic absorption

spectrometry

SPF Spectrofluorimetry CG Gas chromatography

HPLC High performance liquid chromatography

INTRODUCTION

Interest in the role of Se in human nutrition is increasing, as more and more investigators realise the essential nature of this trace element to human health (Levander, 1987). Although interest in Se was initially caused by its potential toxicity (Wilber, 1980), more important is its deficiency in several geographical areas. Low Se intakes have been associated with Keshan and Kashin-Beck diseases, juvenile cardiomyopathies that occur in certain parts of China (Keshan Disease Research Group, 1979; Levander, 1987; Khan, 1989). There are many other diseases that have been related to Se deficiency. Epidemiological studies have shown a relationship between low Se intake and increased risk of cancer (Clark, 1985; Combs & Clark, 1985; Yu et al., 1985; Levander, 1987) or ischemic heart disease (Virtamo et al., 1985; Levander, 1987; Bukkens et al., 1990; Oster & Prellwitz, 1990).

The main source of Se intake are foods. Milk of all animal species is notoriously low in trace elements (Picciano, 1985). Selenium concentration in milk is lower than the concentration of other essential trace elements (Cu or Zn) (Hatano et al., 1985). Breast milk may be the main source of Se for breast-fed infants. Therefore, the section of the population most vulnerable to Se deficiency includes infants receiving breast or formula milk containing very low levels of Se and patients on restricted long-term diets (Lane et al., 1981; Sando, 1989). Selenium requirements of children have been extrapolated from adult values on the basis of body weight, and an arbitrary factor allowed for growth. Thus, the Food and Nutrition Board (US) has proposed a lowest level of safe and adequate intake of Se for infants: 10 and 15 μ g/day for infants of 0-0.5 years and 0.5-1 years, respectively (Committee on Dietary Allowances, 1989).

In this paper, we review the main data published from 1975 to 1992 on Se concentrations in human milk and commercial milk adapted for human consumption. Also, Se intakes for lactating infants from different geographical areas have been included. Given their relevance we have included some data from earlier years.

ANALYTICAL CONSIDERATIONS

Sampling and storage

Milk samples must be collected via a mechanical pump according to the standard procedures described in the IAEA/WHO document (Parr, 1978). All sample collection equipment was plastic or, more specifically,

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polypropylene (Litov et al., 1989), acid-washed to prevent Se contamination. Care should be taken when sampling mature human milk for estimation of Se concentration (Smith et al., 1982) because, although these authors did not find differences in Se content throughout the day, samples should be collected from different feeds during the same day. Also, when possible, various samples were collected during the same feed (Vuori & Kuitunen, 1979; Clemente et al., 1982).

Milk or colostrum samples were freeze-dried (Binnerts, 1979; Varo & Koivistoinen, 1981; Polkowska-Motrenko et al., 1982) or frozen (Maus et al., 1980; Debski et al., 1987; Levander et al., 1987; Mannan & Picciano, 1987) in liquid nitrogen (Smith et al., 1982) or in solid carbon dioxide (Shearer & Hadiimarkos, 1975) immediately after sampling. Then the samples were stored at -14° C (Litov et al., 1989), at -18° C (Karlsen et al., 1981; Kumpulainen et al., 1983b; Varo et al., 1984), at -20° C (Smith et al., 1982), at -70° C (Debski et al., 1987; Mannan & Picciano, 1987), or kept in an ice bath (Smith & Picciano, 1986). Preservation of mature milk at -20°C decreased its GSH-Px activity linearly with time at a rate of 5.8 unit/ml/day (Hojo, 1986a). After thawing, the human milk samples were heated to 40°C and carefully mixed before analysis (Kumpulainen et al., 1983b). Also human and cow's milk and formula samples were dried in silica tubes for 24 h at 70°C and thereafter for 8 h at 100°C (Lombeck et al., 1978).

Sample treatment

For studying the distinct fractions of milk, diverse methods of fractionation have been applied. Skimmed milk has been prepared by centrifugation for 1.5 h at 4°C at 10 000 g (Debski et al., 1987) or for 30 min at 2000 g (Avissar et al., 1991) or at 1500 g (Yoshida et al., 1981; Van Dael & Deelstra, 1989). After being centrifuged for 1 h at 120 000 g (4°C), the supernatant (whey) and pellet (mostly casein) fractions are obtained. After this, the samples have been dialyzed in membrane tubing with a molecular mass cutoff of 6-8 kDa for four days against deionized water (Debski et al., 1987). Also, the pH of the skimmed milk adjusted to 4.6 to precipitate caseins (Van Dael & Deelstra, 1989). CaCl₂ and rennin have been employed in order to precipitate casein at 37°C for 5 h. Afterwards the whey fraction was separated from the casein coagulum by centrifugation at 1000 g for 15 min. Trichloroacetic acid (TCA) was added to the whey fraction and the suspension was subsequently centrifuged to separate it into TCA-soluble and TCAinsoluble fractions. On the other hand, the precipitated casein was washed with NaOH solution, subsequently collected by centrifugation, and dried in a vacuum (Yoshida et al., 1981).

In the case of X-ray fluorescence, neutron activation analysis, and atomic absorption spectrometry with a graphite furnace, sample preparation can be kept to a minimum. So, the treatment of the sample was usually drying (Lombeck *et al.*, 1978) as no differences could

be detected between the drying and ashing methods (Behne & Matamba, 1975). However, most techniques have used digestion as a sample treatment. Selenoorganocompounds in milk can resist oxidation except with HClO₄ mixture (Olson et al., 1975; Nève et al., 1982). However, other authors (Olson et al., 1975; Bunker & Delves, 1987; Charlot et al., 1989) do not find differences in the results when mixtures of HNO, and H₂SO₄ or HNO₃, H₂SO₄ and HClO₄ are added. Other digestion procedures have been proposed, such as the addition of HNO₃, H₃PO₄ and H₂O₂ in order to eliminate the need for a HClO₄ mixture (Reamer & Veillon, 1983; Macpherson et al., 1988). Recently, a somewhat revolutionary method was developed for the digestion of organic and inorganic matrices in biological fluids which involves the use of microwaves (Cornelis, 1991; Feinberg, 1991; LamLeung et al., 1991; Matusiewicz et al., 1991).

Instrumental determination

There are many available methods to determine Se. Four important methods may stand out for the analysis of Se in milk: Instrumental Neutron Activation Analysis (INAA), Spectrofluorimetry (SPF), Gas Chromatography (GC), and Atomic Absorption Spectrometry (AAS).

Spectrofluorimetric methods are well established. They are based on the measurement of the sensitive fluorescence of piazselenols derived from selenite. Thus, the digested samples must be treated with HCl, heating to reduce the selenate to selenite (Lalonde et al., 1982; Koh & Benson, 1983; Pettersson & Olin, 1991). Other reducing agents such as H₂O₂ (Michie et al., 1978) or hydroxylamine (Holynska & Lipinska-Kalita, 1977; Peters & Koehler, 1982) have been used. The optimum pH and temperature/time for the formation of a fluorescent piazselenol with DAN (2,3-diaminonaphthalene) was studied by several authors (Lalonde et al., 1982; Alfthan, 1984; Pettersson et al., 1988). In order to eliminate interferences, this complex was extracted with a hydrophobic solvent such as cyclohexane, toluene, benzene, chloroform, etc. Afterwards, the fluorescence $(\lambda_{\rm ex} = 360 \text{ nm}, \lambda_{\rm em} = 520 \text{ nm})$ was measured in a fluorescence spectrometer (Watkinson, 1966). In biological materials and for concentrations of Se < 0.3 mg/kg fluorimetry has been selected as a reference method, because of its high precision, low detection limit, and complete recovery (Table 1).

New HPLC methods coupled to fluorescence detection of the piazselenol complex are being developed. These methods can be applied to human milk or paediatric samples where small quantities may be available and subnanogram sensitivity is required (Vézina & Bleau, 1988; Handelman et al., 1989). Also, these chromatographic methods show promise because of the fact that the different Se species present in milk could be separated and determined, which would be interesting for bioavailability studies.

INAA, if available, can be valuable as a reference for validating alternative analytical methods. In general,

sensitivity and precision of INAA are lower than fluorimetric methods (Table 1), and both are capable of producing unbiased results (Heydorn & Griepink, 1990). However, INAA has several advantages with respect to fluorimetric methods. Thus, it is non-destructive and the only losses of concern are by escape of volatile compounds. The sample treatment can be reduced to a minimum. Exposure of the sample in a nuclear reactor for a few days yields only one long-lived isotope of Se, ⁷⁵Se. When the short-lived activation product ^{77m}Se was measured the time of analysis decreased significantly from three months to two days (Egan et al., 1977; Woittiez & Nieuwendijk, 1987). Also, this method can be combined with radiochemical separation with orthodiamines in order to eliminate interference (Kalouskova et al., 1989), but the RSD can increase (Sarudi et al., 1989).

Selenium determination in biological fluids in the subnanogram range by AAS can be performed using the graphite furnace (EAAS) and the hydride-generation techniques (HG-AAS) (McCarthy et al., 1981; Ringstad & Thelle, 1986). The two techniques (HG-AAS and EAAS) have been correlated satisfactorily (Oster & Prellwitz, 1982: Macpherson et al., 1988). The best absolute detection limit can be observed in EAAS but HG-AAS is faster and cheaper (Verlinden et al., 1981). An automated microtechnique for Se analysis, flow injection HG-AAS, is capable of performing a rapid and accurate Se determination at picogram levels in acid-digested biological fluids (Negretti de Brätter et al., 1990). The relatively poor precision, losses, and interference are the main problems of both methods. Some authors (Koh & Benson, 1983) have left out the previous digestion step in the EAAS technique. But most authors (Neve et al., 1980; Tôei & Shimoishi, 1981; Koops et al., 1989) prefer to digest the samples of milk, and after carrying out an extraction process in order to eliminate interference. In order to stabilize or reduce the volatility of inorganic Se compounds of digested samples for direct determinations by EAAS, the addition of various metallic salts has been proposed. Salts of nickel are commonly used (Alexander et al., 1980; Carnrick et al., 1983). Also, a Mg-Pd system or Pd alone has a substantial equalizing effect on the atomization temperature (Schlemmer & Welz, 1986; Charlot et al., 1989; Koops et al., 1989). The use of Zeeman-effect background correction is widely recommended in biological matrices where iron and phosphorus are also present (Koops et al., 1989; McMaster et al., 1990; Hoenig, 1991; Welz et al., 1983). In the HG-AAS technique, it is necessary to reduce, with borohydride, the selenate to selenite in the digested samples. Interferring ions in this step can be masked by addition of 1,10-phenanthroline, quinidine-8-ol or thiourea (Long & Yu, 1986).

Methodology available for the quantification of Se by GC was reviewed (Dilli & Sutikno, 1984b). In GC methods, digestion and reduction steps are necessary to determine Se in milk, as in fluorimetric methods. Afterwards the digested and reduced sample is treated with

halogenated aromatic o-diamines to form piazselenols (Uchida et al., 1981; Dilli & Sutikno, 1984a). The selenium complex is extracted in organic solvent and measured by the sensitive GC method with an electron capture detector (Cappon & Smith, 1978; Uchida et al., 1981). However, the sensitivity of GC is not as good as in the latter methods, although the reported precision is lower than those. Also, some authors (Cappon & Smith, 1978) have observed incomplete recovery (Table 1).

SELENIUM IN MILK

Occurrence

The total Se content does not give information on the overall utilization or bioavailability of the element. Selenium in human milk and other animal milks was positively correlated with its protein content (Millar & Sheppard, 1972; Smith et al., 1982; Hojo, 1986b) and negatively correlated with its fat content (Hojo, 1986b). At least 8-12 selenoproteins could be identified after gel chromatography of dialyzed human milk (Debski et al., 1987). Skimmed milk contained 93% of the total Se, which is mainly protein-bound; k-casein was the protein richest in Se, followed by β -casein (Van Dael et al., 1991). Yoshida et al. (1981) have shown a major proportion of Se in the casein fraction of pasteurized bovine milk. The Se in the whey fraction is mainly in the free selenite form. Other investigators (Van Dael & Deelstra, 1989) have found especially, that β -lactoglobulin is a Se-rich protein, contributing up to 80% of the total Se content of the bovine whey. Separation treatment influences the Se content in each fraction obtained. So, only 1-3% of total milk Se remained in the lipid fraction after centrifugation. Approximately 20-28% of Se in milk was removed by dialysis but the loss of Se from dialysis was not uniform among fractions; there was a loss of 66% of the Se associated with the 10 kDa fraction in human milk. After ultracentrifugation, the supernatant fractions contained 62, 71, and 29% of the total milk Se collected from women, cows, and goats, respectively (Debski et al., 1987). Consequently, the pellet obtained from goat's milk contained 65% more Se than the pellet from human milk and 204% more than that from cow's milk (Debski et al.,

The strong positive relationship (r = 0.81, p < 0.001) observed between Se concentration and GSH-Px activity of human milk suggests that a large portion of the Se in human milk is present as a part of this enzyme (Mannan & Picciano, 1987). This was verified by molecular sieve chromatography of human milk with GSH-Px and accounted for 15–30% of the Se in milk (Milner et al., 1987). In cow's milk about 12% of Se was bound to GSH-Px (Hojo, 1982). Most of the GSH-Px activity was found in the fractions corresponding to 170 kDa and 96 kDa in milk from women, goat, and cow species examined (Milner et al., 1987). There are two distinct forms of Se-dependent gluta-

Table 1. Literature data on determination of Se in milks

radic 1. Literature data on determination of Se in mins	Sample treatment Detection RSD % Recovery Ref. limit between-assay % (within-assay)	digestion; HCl reduction; DAN; 0.2 ng Grant (1981)	digestion; HCl reduction; 10 100 ± 2.2 Koh & Benson (1983) cyclohexane extraction (2) (97–100)	digestion; HCl reduction; 5 Koops et al. (1989) cyclohexane extraction	H_2SO_4/H_2O_2 digestion; NH ₂ OH reduction; 50 4·0 102–107 Nêve <i>et al.</i> (1980) aminobenzene; toluene extraction; Ni(NO ₃) ₂	H ₂ SO ₄ /H ₂ O ₂ digestion; HCl reduction; 0.6 7.3 98.4 Kumputainen <i>et al.</i> (1983 <i>a</i>) omplexation; APDC-MIBK extraction	H ₂ SO ₄ /H ₂ O ₂ digestion; HCl reduction; 93·1 dIBK extraction	H ₂ SO ₄ /H ₂ O ₂ digestion; HCl reduction; omplexation; APDC-MIBK extraction	H ₂ SO ₄ /H ₂ O ₂ digestion; HCl reduction; omplexation; NaDDC-MIBK extraction	H ₂ SO ₄ digestion; HCl reduction; -MIBK extraction; Cu ²⁺ modifier	digestion; HCl reduction; DAN, cyclohexane 0.5 2 Norheim et al. (1983) inic Ag-sulfonate/hydrocarbon oil	digestion, urea; 20 Charlot <i>et al.</i> (1989) PdCl ₂ as matrix modifier	lilution; Pd/Mg(NO ₃₎₂ as matrix modifier 18 Koops <i>et al.</i> (1989)	stion/silicic acid; 2 ng 1·6 Han et al. (1981) scid medium; NaBH ₄	
	Sample treatment	HNO ₃ /HClO ₄ digestion; HCl reduction; DAN; cyclohexane extraction	HNO ₃ /HClO ₄ digestion; HCl reduction; EDfA, DAN, cyclohexane extraction	HNO ₃ /HClO ₄ digestion; HCl reduction; EDTA, DAN, cyclohexane extraction	$HNO_3/HCIO_4/H_2SO_4/H_2O_2$ digestion; NH_2OH reduction; 4-chloro-1,2-diaminobenzene; toluene extraction; $Ni(NO_3)_2$ modifier	HNO ₃ /HClO ₄ /H ₂ SO ₄ /H ₂ O ₂ digestion; HCl reduction; EDTA, Cu ²⁺ complexation; APDC-MIBK extraction	HNO ₃ /HClO ₄ /H ₂ SO ₄ /H ₂ O ₂ digestion; HCl reduction; Cu ²⁺ ; APDC-MIBK extraction	HNO ₃ /HClO ₄ /H ₂ SO ₄ /H ₂ O ₂ digestion; HCl reduction; EDTA, Ni ²⁺ complexation; APDC-MIBK extraction	HNO ₃ /HClO ₄ /H ₂ SO ₄ /H ₂ O ₂ digestion; HCl reduction; EDTA, Cu ²⁺ complexation; NaDDC-MIBK extraction	HNO ₃ /HCIO ₄ /H ₂ SO ₄ digestion; HCI reduction; EDTA, APDC-MIBK extraction; Cu ²⁺ modifier	HNO ₃ /HClO ₄ digestion; HCl reduction; DAN, cyclohexane extraction organic Ag-sulfonate/hydrocarbon oil	HNO ₃ /H ₂ SO ₄ digestion, urea; Triton X-100; PdCl ₂ as matrix modifier	Triton X-100 dilution; Pd/Mg(NO ₃) ₂ as matrix modifier	Oxygen combustion/silicic acid; dissolution in acid medium; NaBH ₄	HNO,/HClO,/H,SO, digestion: HCl reduction: NaBH,
	Method	SPF	SPF	SPF	EAAS	EAAS with D ₂ lamp				EAAS with D_2 lamp	EAAS	EAAS with D_2 lamp	EAAS, Zeeman	HG-AAS	HG-AAS with

Lombeck et al. (1977); Lombeck et al. (1980)	Clemente et al. (1982)	Woittiez & Nieuwendijk (1987)	Polkowska-Motrenko et al. (1982)	Singh & Sawant (1987)	Shimoishi (1976) Tôci & Shimoishi (1981)	Stijve & Philippossian (1978)	Cappon & Smith (1978)	Dilli & Sutikno (1984a)
					94-106		75–90	
10	6.7	15		20	1.7		3.4	
	l ng/g	2 ng		at.	S ng	5 ng/g	0.19 ng/g	S
Dried at 70°C, 24 h and at 100 °C, 8 h; $5 \cdot 10^{13}$ n/cm ² · s, 48 h irradiation; 60 –90 days of decay time	$2.6 \cdot 10^{12} \text{ n/cm}^2 \cdot \text{s}$, $10-14 \text{ h}$ irradiation	Lyophilization; $5 \cdot 10^{13} \text{ n/cm}^2 \cdot \text{s}$, 2 s irradiation; 5 s decay time; 10 s counting time	Lyophilization; oxygen flask combustion; HCVH ₂ SO ₄ digestion; DDTC/toluene extraction; 2 · 10 ¹² n/cm ² · s, 20–40 h irradiation	Dried at 65°C; ethyl- α -isonitroso-acetoacetate precipit. HNO ₃ /HClO ₄ redissolution; $1 \cdot 10^{13} \text{ n/cm}^2 \cdot \text{s}$, 5–7 days irradiat.	HNO ₃ digestion; urea; HCl reduction; toluene extraction; 1,2-diamino-4-nitrobenzene or 1,2-diamino-3,5-dibromobenzene; toluene extraction	HNO ₃ /Mg(NO ₃) ₂ digestion; HCl reduction; urea; 1,2-diamino-4-nitrobenzene; toluene extraction	HNO ₃ digestion; urea; HCl reduction; 4-nitro-o-phenylenediamine; benzene extraction; MgSO ₄ /Florisil cleanup	HNO ₃ /HClO ₄ digestion; HCl reduction; 4-trifluoromethyl-o-phenylenediamine; toluene extraction
INAA, Se ⁷⁵	INAA, Se ⁷⁵	INAA, Se ⁷⁵ , Se ^{77m}	RNAA. Se ⁷⁵	RNAA, Se ⁷⁵	GC-ECD	GC-ECD	GC-ECD	GC-ECD

thione peroxidase in mammals, a cellular form (c-GSHPx) and an extracellular or plasma form (p-GSHPx) (Broderick et al., 1987; Maddipati & Marnett, 1987; Takahashi et al., 1987; Avissar et al., 1989a,b). The two enzymes have different affinities for glutathione and hydroperoxides. Both p-GSHPx and c-GSHPx have four Se molecules, interact with lipid peroxides and H₂O₂, and are the products of two different genes (Takahashi et al., 1990; Avissar et al., 1991). Thus, most, if not all, GSH-Px activity in milk is due to the p-GSH-Px form of the enzyme (Avissar et al., 1991).

Levels in human milk

Table 2 shows the mean and range of Se concentrations in human milk from different parts of the world, also indicating, the number of samples analyzed, analytical method, some important descriptions and approximate year of the study. Most authors determine total Se levels because the normal sample treatment changes the initial composition of the milk. Studies in human milk from Japan (Shimoishi, 1976; Tôei & Shimoishi, 1981) determine the Se (VI) species, indicating that they constitute 30% of the total Se. The unit of concentration most used is $\mu g/litre$, although others have been used, such as pmol/g (Mangels et al., 1990), ng/litre (Higashi et al., 1983), μg/ml (Shimoishi, 1976; Tôei & Shimoishi, 1981), ng/g wet weight (Binnerts, 1979; Clemente et al., 1982; Robberecht et al., 1985; Tamari et al., 1990, 1991) or dry weight (Lombeck et al., 1977, 1978; Binnerts, 1979; Polkowska-Motrenko et al., 1982; Kumpulainen et al., 1983a), µg/g wet weight (Norheim et al., 1983), µmol/litre (Koh & Benson, 1983). However, all concentrations were converted to $\mu g/litre$ by using a density of 1.034 (Iyengar, 1982) or a mean dryto-wet ratio of 0.12 (Cross et al., 1978; Lombeck et al., 1978; Iyengar, 1982).

Mature human milk can exhibit a wide variation in Se content due to geographical location. Shearer and Hadjimarkos (1975) have carried out an important study in 17 states across the USA, observing significantly higher concentration levels in high Se areas such as South Dakota compared to lower Se areas such as Ohio. Also, it was found that breast milk in a high Se area contained 283 µg Se/kg which can provide 220 µg Se/day for infants within the first year of age and is around 90 times the Se-intake supplied by breast milk in the Keshan disease area, where the Se concentration averaged 2.6 µg Se/kg (Keshan Disease Research Group, 1979a,b; Yang et al., 1989). In 26 lactating women and their infants from the Kathmandu valley of Nepal, Se daily intake is considered the second-lowest reported in the world. However, the concentrations in maternal and infant plasma, and breast milk are below normal (Reynolds et al., 1986). Kumpulainen et al. (1983b, 1984) suggest that maternal dietary Se intakes may be insufficient to maintain breast-milk Se concentrations at adequate levels for infants because Finland is a country with low-Se soils. Although some authors

(Levander et al., 1981, 1987; Higashi et al., 1983) did not find significant correlations between Se levels in breast milk and dietary Se intakes, others (Shearer & Hadjimarkos, 1975; Kumpulainen et al., 1985; Yang et al., 1989) claimed that the Se content in human milk reflected a different Se intake via food.

Most authors indicate that human milk Se concentrations are dependent on the maternal Se intake; consequently, Se in human milk must also be dependent on the Se status. Thus, correlational analyses were performed between maternal indices of Se status and milk indices. Strong positive relationships were observed between maternal plasma (Levander et al., 1981; Mannan & Picciano, 1987), serum (Higashi et al., 1983; Kumpulainen et al., 1985) or whole blood (Williams, 1983) Se concentrations and Se in milk. Plasma activity of GSH-Px was also positively correlated with both milk Se concentration and activity of GSH-Px (Mannan & Picciano, 1987). However, erythrocyte Se content did not correlate with milk Se concentrations (r = 0.23, p:NS) (Mannan & Picciano, 1987).

Some authors (Levander et al., 1987) indicated that there was a weak (r = 0.38) but statistically significant (p < 0.025) correlation between maternal plasma and milk. In contrast, other investigators found that the level of Se in serum (Higashi et al., 1983) or plasma (Levander et al., 1981) of lactating women was unrelated to the Se content of their milk. The ability to demonstrate a correlation between blood and breastmilk Se levels may be determined in part, by the form of dietary Se ingested by the mothers (Levander et al., 1987).

Another important factor that seems to influence Se content in milk is the state of lactation. Thus, the mean content found in USA (18 μ g/litre) was lower than in North-West Germany (28 μ g/litre). This could be due to the different state of lactation (Lombeck et al., 1978) and/or different dietary Se intake. The 241 American women donating milk samples were from 17 to 869 days postpartum, with a median duration of 183 days, while most (88%) of the samples of mature milk from German women were collected from 11 to 60 days postpartum because of the usually short lactation time in this area (Lombeck et al., 1978). Mature human milk exhibited a Se content very much lower than that observed in colostrum milk (Grimanis et al., 1978; Lombeck et al., 1978; Smith et al., 1982; Higashi et al., 1983; Robberecht et al., 1985; Tamari et al., 1991). Also, the available data reveal a strong downward trend in breast milk Se concentration during the first 10 days of lactation (Millar & Sheppard, 1972; Grimanis et al., 1978; Higashi et al., 1983; Varo et al., 1984; Tamari et al., 1990, 1991, 1992; Yuzo & Mohri, 1991). Cumming et al. (1983) showed that, in a group of lactating Australian women, the mean plasma/milk selenium ratio was 6.9 ± 1.83. This ratio increases slightly with the progression of lactation; while the plasma concentration remains constant the milk concentration declines. Lombeck et al. (1978) have shown an inverse relationship between the Se content and the

Table 2. Selenium levels (μ g/litre) in human milk

Geographical location	Date (year)	Anal. $method^a$	Mean (μg/litre) (min-max)	Nº.	Description ^b	Ref.
EUROPE			· · · · · · · · · · · · · · · · · · ·			
Austria (Styria)	198891	HG-AAS	8·3 ± 3·0	34	Mature milk	Tiran et al. (1992)
Belgium	1983	HG-AAS	14.3 ± 4.7 (6.4–26.3)	24	Colostrum 0–3 days pp	Robberecht et al. (1985)
			11.9 ± 4.2	13	Transitory milk	
			(5.3-18.1) 12.3 ± 3.4	11	5-7 days pp Transitory milk	
			(6.5-17.1)	11	8–10 days pp	
			$9.1 \pm 1.9^{\circ}$	15	Mature milk	
			(6.5-12.3)	9	l month pp Mature milk	
			9.6 ± 3.3 (5.4–14.9)	9	2 months pp	
			9.3	1	Pooled milk	
Finland	1976	EAAS	10.7 ± 1.6	13	1 month pp	Kumpulainen et al. (1983b)
(Helsinki)			5.8 ± 1.2	13	3 months pp	-
F21 1 1	1000	E 1 1 0	5.6 ± 0.4	4	6 months pp	TZ 1 1
Finland (Helsinki)	1980	EAAS	11·8 10·0	_	1 month pp 3 months pp	Kumpulainen et al. (1984)
Finland	1982	EAAS	6.0 ± 0.5		Dried milk	Kumpulainen et al. (1983a)
Germany (Düsseldorf)	1976	INAA	14·9 (11·4–18)	3	≥15 days pp	Lombeck et al. (1977)
Germany	1978	INAA	77.9	3	Colostrum	Lombeck et al. (1978)
(Düsseldorf)			(60.8-100.4)		2-3 days pp	
			28.4	25.	Transitory milk	
			(15·1–48·9) 26·7	44	4-10 days pp Mature milk	
			(10.4-50.1)		11–173 days pp	
Germany	1986	SPF	17.8	_	10 days pp	Oster et al. (1986)
Greece (Athens)	1973	SPF	21 ± 1	24	29-195 days pp	Hadjimarkos & Shearer (1973)
Greece	1978	INAA	48	15	Colostrum	Grimanis et al. (1978)
			(33–69) 16	15	0-3 days pp Transitory milk	
			(10–20)	15	4–10 days pp	
			15	5	Mature milk	
			(14–22)		1 month pp	
Italy	1982	INAA	$12.9 \pm 0.9 (\le 0.9 - 47.9)$	20	≥15 days pp	Clemente et al. (1982)
Spain (Barcelona)	1981	SPF	11.4 ± 2.9 (9–18)	8	3-10 days pp	Farré et al. (1981)
UK (Glasgow)	1978		31 ± 10	25	_	Cross et al. (1978)
Yugoslavia (Ljubljana)	1982	RNAA	(7-4-14-5)	10		Polkowska-Motrenko et al. (1982)
Yugoslavia (Ljubljana)	1983	RNAA	$ \begin{array}{r} 11.5 \pm 3.6 \\ (5.7 - 16.7) \end{array} $	27	0-14 days pp	Kosta et al. (1983)
AMERICA USA (Arizona,	1974	SPF	20 ± 1	14	22-344 days pp	Shearer & Hadjimarkos (1975)
Douglas)	1217		(11–27)		18-32 yr old	······································
USA (Calif.,		SPF	18 ± 2	14	28–360 days pp	
Sta Barbara) USA (Colorado,		SPF	(10–32) 15 ± 1	15	22–35 yr old 86–648 days pp	
Pueblo)		V. I	(12–20)		19-33 yr old	
USA (Connect.,		SPF	15 ± 1	15	29-425 days pp	
Bristol) USA (Georgia,		SPF	$(8-23)$ 18 ± 1	11	19–29 yr old 28–360 days pp	
A II -DATEMIC						

Table 2—continued

Geographical location	Date (year)	Anal. method ^a	Mean (μg/litre) (min-max)	Nº.	Description ^b	Ref.
IMERICA—contd.						
USA (Illinois)	1981	GC-ECD	41.2 ± 17.3	8	Colostrum 0–3 days pp	Smith et al. (1982)
(11111015)			18.0 ± 3.8	8	1 month pp	
			15.7 ± 4.6	8	2 months pp	
			15.1 ± 5.8	8	3 months pp	
			15·7 ± 4·9	20	Fore milk	
			13/ ± 47	20	2 weeks pp	
			14·4 ± 1·6	20	Fore milk	
			144110	20	1 month pp	
			14.1 ± 3.2	20	Fore milk	
			14.1 ± 3.2	20	2 months pp	
			13.9 ± 3.2	20	Fore milk	
			13.9 ± 3.2	20		
			162140	70	3 months pp	
			16.3 ± 4.9	72	Average	
			(8–34)	20	fore milk	
			16.3 ± 4.8	20	Hind milk	
			150.00	••	2 weeks pp	
			15.2 ± 3.2	20	Hind milk	
					1 month pp	
			15.9 ± 3.4	20	Hind milk	
					2 months pp	
			16.4 ± 3.1	20	Hind milk	
					3 months pp	
			16.2 ± 5.4	16	Morning milk	
			16.6 ± 4.8	16	Midday milk	
			15.8 ± 4.6	16	Evening milk	
USA	1984–85	GC-ECD	16.8	10	4-8 weeks pp	Mannan & Picciano (1987)
(Illinois)			(11.5-27.3)		30 ± 5.6 years old	
			15.6 ± 0.4	10	Fore milk	
			18.1 ± 0.6	10	Hind milk	
USA	1986	GC-ECD	15.2 ± 0.6	10	_	Debski et al. (1987)
(Illinois)			15.1 ± 0.9	10	Skimmed milk	,
,			11.0 ± 0.5	10	Dialyzed milk	
USA (Indiana, Evansville)	1989	SPF	23 ± 4	12	0-2 months pp	Litov et al. (1989)
USA (Iowa,	1973	SPF	20 ± 1	15	40-83 days pp	Hadjimarkos & Shearer (1973
Iowa City)			(15–24)		18–31 yr old	-
USA	1981		20 ± 4	23	1 month pp	Levander et al. (1981)
(Maryland)	.,,,		15 ± 3	23	3 months pp	201411401 01 41. (1301)
(1.141)14114)			15 ± 4	23	6 months pp	
USA	1986	SPF	20 ± 1	10	1 months pp	Levander et al. (1987)
(Maryland)	1700		15 ± 1	10	3 months pp	
(Arama jamanu)			15 ± 1	10	6 months pp	
USA (Missouri,	1974	SPF	20 ± 1	15	22-540 days pp	Shearer & Hadjimarkos (1975
Rolla)	17/7	OI I	(13–28)	13	22–340 days pp 22–37 yr old	Shearer & Haujimarkos (1973
USA (Montana,			$(13-28)$ 21 ± 1	10	42–37 yr old 42–390 days pp	
				10		
Billings)			(16–27)	1.5	24–30 yr old	
USA (New York,			15 ± 1	15	23-869 days pp	
Syracuse)	1000		(10–19)		25–39 yr old	A
USA	1980	_	56		Colostrum	Amin et al. (1980)
(New York)	1000 00	CDE	31	_	Colostrum	
USA (New York,	1989–90	SPF	13.1	3	1-8 months pp	Avissar et al. (1991)
Rochester)						
USA (Ohio,	1974	SPF	13 ± 1	15	20-353 days pp	Shearer & Hadjimarkos (1975
Akron)			(7–17)		23–42 yr old	
USA (Oklahoma,			16 ± 1	14	43-630 days pp	
Norman)			(9–26)		20-36 yr old	
USA (Oregon,	1973	SPF	21 ± 3	15	42-150 days pp	Hadjimarkos & Shearer (1973
Portland)			(13-53)		17–44 yr old	

Table 2—continued

Geographical location	Date (year)	Anal. method ^a	Mean (μg/litre) (min-max)	Nº.	Description ^b	Ref.
AMERICA—contd.						
USA	1974	SPF	21 ± 1	15	65-363 days pp	Shearer & Hadjimarkos (1975
(Pennsylvania,			(12–26)		22-33 yr old	
State College)			, ,		·	
USA			28 ± 3	15	112–389 days pp	
(South Dakota,			(17–60)		21–41 yr old	
Sioux Falls)						
USA (Texas,			16 ± 1	15	24-480 days pp	
Corpus Christi)			(9–22)		20-31 yr old	
USA (Utah,			22 ± 3	14	17-390 days pp	
Salt Lake City)			(14–52)		21–32 yr old	
USA (Utah,	1989	SPF	25 ± 5	10	0–2 months pp	Litov et al. (1989)
Salt Lake City)						
USA	1989	IDMS	204 ± 13	6	74 ± 5 days pp	Mangels et al. (1990)
(Washington DC			(151-327)		29 ± 2 yr old	
USA (Wyoming,	1974	SPF	16 ± 1	14	37-704 days pp	Shearer & Hadjimarkos (197)
Cheyenne)			(10–26)		17–38 yr old	
4SIA						
Japan	1975-80	GC-ECD	45 ± 38	3		Robbins & Caruso (1979);
(Okayama)			(10-83)			Shimoishi (1976);
(,			,			Tôei & Shimoishi (1981)
Japan (Kyoto)	1982		22.5 ± 4.2	13	Mature milk	Hojo (1982)
Japan	1983	SPF	(35-152)	7	Colostrum	Higashi <i>et al.</i> (1983)
1			,		(0-3 days pp)	
			(15-79)	10	Transitional milk	
			, ,		(4–10 days pp)	
			(9-39)	9	Mature milk	
					(1 month pp)	
			(6–28)	8	(3 months pp)	
			(9–33)	7	(5 months pp)	
			21	34	Average content	
					without colostrum	
Japan	1986	SPF	22.5	_	Mature milk	Hojo (1986b)
Japan	1990	SPF	19.6		Colostrum	Tamari <i>et al.</i> (1990)
					(2 days pp)	
			0.8	_	Milk, 5 days pp	
Japan	1991	SPF	28.4	_	Colostrum	Tamari et al. (1991, 1992)
			8.7		Mature milk	
				_	(7–171 days pp)	
Japan	1991	SPF	29.7 ± 10.5	8	Colostrum	Yuzo & Mohri (1991)
			18.9 ± 9.4	8	Transitional milk	
				_	(4-7 days pp)	
			10.8 ± 2.7	8	Mature milk	
					(7–21 days pp)	
OCEANIA						
Australia	1982	INAA	11.6 ± 1.9*	14	8-23 weeks pp	Cumming et al. (1983)
			(9.7-29)		• •	` ` ` '
Kensington	1984	GC-ECD	8	l	Healthy	Dilli & Sutikno (1984a)
New Zealand	1983		7.6		1 month pp	Williams (1983)

^a GC-ECD: Gas-chromatography with electron capture detection; IDMS: Isotope dilution mass spectrometry.

time postpartum; however, other authors (Shearer & Hadjimarkos, 1975) have indicated that this correlation was very weak (r = -0.13, p < 0.05). Another author (Hojo, 1986a) showed that both GSH-Px and Se contents of breast milk decreased with increasing time of lactation and reached a plateau one month postpartum. This sequential change was not due to the Se intake of the mothers, as reflected in urinary Se content. This is

in accordance with the values in serum (Lombeck et al., 1978; Kumpulainen et al., 1987) and whole blood (Mackenzie et al., 1978) Se concentrations of children. Also a strong decrease in breast-milk Se during the first and third months of lactation has been observed in lactating women from Finland ($\approx 30~\mu g$ Se/day) (Kumpulainen et al., 1983b) and Belgium (Robberecht et al., 1985). However, in mature human milk from the

^b pp: Postpartum.

^{*}pmol Se/g.

Table 3. Selenium levels (μg /litre) in cow's milk

Geographical location	Date (year)	Anal. $method^a$	Mean (min-max)	Nº.	Description	Ref.
EUROPE					The work of the state of the st	also area recommendation of the contraction of the
Finland	1977	HG-AAS	7·89 ± 1·2	16	Dairy cows	Varo & Koivistoinen (1981)
	1977		3.48	6	Dairy cows	, a. c & c
			(3.48 - 4.64)		(standardized)	
	1980		8-12	6	Dairy cows	
			(6.96-9.28)		(untreated)	
Germany	1976	INAA	10·2 ± 2·55	3	Dairy cows	Lombeck et al. (1977)
(Düsseldorf)			(7.8-12.9)		- ···· , · · · · ·	
Germany	1978	INAA	23.2	45	Dairy cows	Lombeck et al. (1978)
(Düsseldorf)			$(16\cdot 2 - 35\cdot 2)$			
Germany	1986	_	(6.8-7.2)		Dairy cows	Oster et al. (1986)
Norway	1979	SPF	10.6	13	Dairy cows	Karlsen et al. (1981)
(Oslo)			(7.7-11.6)		•	, ,
Norway	1983	EAAS	10.1 ± 2.5	10	Cows	Norheim et al. (1983)
-			$(7\cdot 2-15\cdot 3)$, ,
The Netherlands	1973	INAA	7.6 ± 0.82	5	Soil type	Binnerts (1979)
			(7.5-13.3)		marine clay	\ ,
			3.9 ± 0.30	5	Soil type	
			$(3\cdot 4 - 5\cdot 45)$		sand and peat	
			5.1 ± 0.63	6	Soil type	
ring and the t	1000	CDE	(5.2–7.54)	~	mixed	T/
The Netherlands	1989	SPF	16.5 ± 1.3	7	Winter milk	Koops et al. (1989)
		EAAS	(14·5–18·4) 15·5	7		
		EAAS	(14·5–17·4)	7		
		SPF	(14.3-17.4) 10.3 ± 0.5	7	Summer milk	
		J. 1	(9.7–10.6)	,	Summer mink	
		EAAS	9.7	7		
			(7-7-13-5)			
<i>MERICA</i>						
Canada	1979	SPF	28	8	Holstein cows	Fisher et al. (1980)
(British Colum	bia)					(· · · · · /
USA (Florida)	1980		(8–13)		Beef cattle	Ammerman at al. (1000)
oba (Fiorida)	1700		(0-13)	_	Se supplement.	Ammerman et al. (1980)
USA (Illinois)	1987	GC-ECD	9·6 ± 0·4	10	Holstein cows	Debski et al. (1987)
(2)				• •	(whole milk)	= toom or an (1701)
			9.5 ± 0.3	10	Holstein cows	
					(skimmed milk)	
			7.3 ± 0.3	10	Holstein cows	
					(dialyzed milk)	
USA (Indiana)	1977	_	(14–23)		Hereford cows	Perry et al. (1977)
					Se supplement	
					2-3 days	
			(16, 21)		postpartum Hereford cows	
			(16–21)		Se supplement	
					3 months	
					postpartum	
USA (Ohio)	1978	SPF	8 ± 1.9	5	Jersey and	Conrad & Moxon (1979)
					Holstein cows	(22.72)
USA (Ohio)	1981		5		Beef cattle	Moxon (1981)
USA	1977	SPF	64 ± 13	8	Dairy cows	Olson & Palmer (1984)
(South Dakota)			(45–80)	^	(December)	
	1978		53 ± 20	8	Dairy cows	
			(32–88)	n	(January)	
			46 ± 7	8	Dairy cows	
			(36–58) 64 ± 10	0	(May)	
			(54–80)	8	Dairy cows (September)	
			1.75***(01/)		COCOCCIIIOCLI	
			78 ± 27	8	Dairy cows	

Table 3—continued

Geographical location	Date (year)	Anal. method ^a	Mean (min-max)	Nº.	Description	Ref.
ASIA						
India (Bombay)	1987	INAA	1.5 ± 3	-	Dairy cows	Singh & Sawant (1987)
Japan	1980	GC-ECD	$(21-27) \text{ Se}_{\text{T}}$	3	Dairy cows	Tôei & Shimoishi (1981)
			(12-17) Se (VI)		-	
Japan	1981		64		Dairy cows	Munnehiro et al. (1981)
Japan (Kyoto)	1982	SPF	23	_	Dairy cows	Hojo (1982)
Japan	1990	SPF	17.4 ± 3.6	13		Tamari et al. (1990)
OCEANIA						
Australia	1982	SPF	$0.2 \pm 0.01*$	4	Raw milk	Koh & Benson (1983)
New Zealand	1980	SPF	7-2	_		Grant (1981)

[&]quot; GC-ECD: Gas-chromatography with electron capture detection.

USA (Shearer & Hadjimarkos, 1975; Smith et al., 1982) or Japan (Higashi et al., 1983), with an average dietary intake of 80 and 88 μ g/day respectively, the concentration of Se does not usually decline significantly with the advancing stages of lactation. Maternal body reserves and/or low dietary Se intake may not have been sufficient to maintain the Se level of breast milk (Kumpulainen et al., 1983b).

Variations in Se concentrations of human milk can occur during the same feed. Se content in hind milk was found to be greater than fore milk at two and three months postpartum (Smith $et\ al.$, 1982) and in the first four months also (Mannan & Picciano, 1987). However, no significant differences were found in human milk samples of ≤ 1 month taken before, during and after feeding (Millar & Sheppard, 1972; Smith $et\ al.$, 1982). Maternal stature, maternal weight or infant's birth weight contribute to the variation in elemental concentration in milk although the contributions were small in all cases (Yoshinaga $et\ al.$, 1991). There was no correlation between Se levels and the ages of the individual donors (Shearer & Hadjimarkos, 1975).

Levels in other commercial milk adapted to human milk

Selenium concentration in milk from different animal species decreases in the order: human, sheep > goat > cow (Debski et al., 1987). The Se contents in commercial milks adapted to human milk, or to cow's milk and other derivatives have been reported showing the results in ng/g dw (Lombeck et al., 1977, 1978), µg/litre (Shimoishi, 1976; Farré et al., 1981; Hojo, 1982), μg/kg (Bruhn & Franke, 1977; Charlot et al., 1989), or ng Se/kcal (Roekens et al., 1985). Although there are exceptions (Higashi et al., 1982; Hatano et al., 1985), most authors (Lombeck et al., 1977, 1978; Farré et al., 1981; Smith et al., 1982; Hojo, 1986; Debski et al., 1987; Tamari et al., 1991, 1992) reported that the Se content of human milk is significantly greater than in the available milk formulas for infants. In this sense, the Se content of infant formulas marketed in the USA

is approximately 40% of that of mature human milk (Smith et al., 1982). Also, the average Se content of 107 samples of 10 different cow's milk infant formulas from Germany was less than 33% of that of mature human milk (Lombeck et al., 1978). However, several authors (Lombeck et al., 1977; Tamari et al., 1990; Hojo, 1982) did not find significant differences between the values of mature human milk and of cow's milk (Table 3). Due to the fact that many Se compounds in milk are quite volatile, processing by hot-air treatment could reduce the Se content of cow's milk with increasing temperature and time of heating (Morris & Levander, 1970; Lombeck et al., 1977). So, Hojo (1986b) found losses of Se amounted to 11.1% at 210°C for 25 min. On the other hand, mean levels of Se (µg/litre) decreased in the sequence: pasteurized cow's milk (28.4) > raw cow's milk (23.1) > mature human milk (22.5) > milk-based infant formula (6.6) (Hojo, 1986b). Lombeck et al. (1980) estimated the Se content of samples of raw cow's milk, ultra-high temperature sterilized milk, and skimmed milk. There was no statistically significant differences in comparison with the respective pasteurized milk samples from the same geographic region. Absence of GSH-Px activity in pasteurized milk and infant formula results from the heating process in their preparation (Hojo, 1986b). In one study (Zabel et al., 1978), infant formulas, not generally supplemented with Se, varied widely in Se content, depending upon their source of ingredients. Also, Se content of nonfat dry milk varies greatly depending on the geographic origin of the milk (Varo et al., 1984). According to the latter, the greater Se content in cow's milk was found in South Dakota, a typical seleniferous area (Olson & Palmer, 1984). On the other hand, the areas which correspond with low Se contents in milk may in fact be Se-deficient (Binnerts, 1979).

There are other factors that can produce some variations in the Se content of cow's milk. Thus, untreated cow's milk obtained in July contained higher levels of Se than that obtained in November (Hojo, 1982). However, in processing plants in South Dakota state, the

^{*} µmol Se/litre.

Table 4. Estimated daily intake of selenium ($\mu g/day$) for infants

Geographical location	Data (year)	Subjects	Feed type	Mean (min-max)	Ref.
UROPE					
Belgium	1983	Boys,	Breast milk	8.1	Tiran et al. (1992)
		3 months			
		Girls,		7 ·1	
Belgium	1984	3 months 1 month	Breast milk	6.1	Roekens et al. (1985)
Deigium	1704	1 monin	Dieast iiiik	(3.8–10.4)	Rockells et at. (1965)
		3 months		7.2	
				(4.5-12.4)	
		6 months		8.6	
				(5.4-14.8)	
		3 months	Bottle-fed cow's milk	9.0	
		1 month	Milk-based formula	3.0	
		2		(0.4–10.9)	
		3 months		3.5	
		6 months		(0·5–12·9) 4·2	
		o months		(0.6-15.4)	
		1 month	Milk infant formula	5.7	
			therapeutic use	(0.6-15.9)	
		3 months		6.7	
				(0.7-18.9)	
		6 months		8-0	
				(0.8-22.5)	
		1 month	Processed cow's milk	7.6	
		2 .1		(6.1-12.8)	
		3 months		9.0	
		6 months		(7·2–15·2) 10·8	
		o montus		(8.6–18.1)	
				,	
Finland	1976	1 month,	Breast milk	8.0 ± 1.8	Kumpulainen et al. (1983b)
(Helsinki)		n = 10	•		
		3 months,		4.7 ± 1.1	
Finland	1981-82	n = 10 4–12 months	Milk infant formula	16	Example and A (1007)
(Helsinki)	1901-02	n = 16	(Se supplemented)	10	Kumpulainen et al. (1987)
(Heisinki)		11 – 10	(se supplemented)		
Germany	1976–77	2 months	Cow's milk	7.8	Lombeck et al. (1978)
(Düsseldorf)			infant formula		
~	1978	2 months	Breast milk	22.4	
Germany	1986	3.5 kg	Breast milk	12	Oster et al. (1986)
		of weight	Milk infant formula Cow's milk	< 3.5	
			COW S IIIIK	5	
Spain	1981	1 week	Breast milk	2.39	Farré et al. (1981)
(Barcelona)		1st week	Milk infant formula	2.34	
		2nd week	Breast milk	3.19	
			Milk infant formula	3.13	
		3–4 weeks	Breast milk	3.19	
		and manch	Milk infant formula	3.90	
		2nd month	Breast milk Milk infant formula	3·23 6·25	
		3rd month	Breast milk	6·25 4·18	
		ora monun	Milk infant formula	8.65	
		4th month	Breast milk	4·53	
			Milk infant formula	9.96	
			Bottle-fed cow's milk	6.94	
		5–6 months	Breast milk	4.53	
			Milk infant formula	10.27	
			Bottle-fed cow's milk	6⋅78	
		7–9 months	Breast milk		
			Milk infant formula	9·66	
			Bottle-fed cow's milk	6.82	
		10 12 manek-	Dragat mills		
		10-12 months	Breast milk Milk infant formula	 7·33	

Table 4—continued

Geographical location	Data (year)	Subjects	Feed type	Mean (min-max)	Ref.
EUROPE—contd.		-			
UK (Scotland)	1978	3 months	Breast milk Milk infant formula	35 18	Cross et al. (1978)
AMERICA					
USA	1978	0-6 months	Milk based diets	8.5	Zabel et al. (1978)
USA (Illinois)	1981	3 months	Breast milk Milk infant formula	10.08 ± 2.96 7.22 ± 1.26	Smith <i>et al.</i> (1982)
USA (Maryland)	1986	1 month 3 months 6 months	Breast milk	10 12 13	Levander et al. (1987)
	1990	8–12 months n = 26	Formula-fed Cow's milk	31 34	Gropper et al. (1990)
4SIA					
China	1985–86	_	Breast milk	11.8	Yang et al. (1989)
(low Se area)					
(medium Se area)				24.3	
(high Se area)		_		94.6	
Japan	1983	5 months	Breast milk	17	Higashi et al. (1983)
Japan	1991	0-4 days	Colostrum	17	Tamari et al. (1991, 1992)
•		>1 month	Breast milk	8	
Japan	1991		Formula-fed	(2-3)	Tamari et al. (1991)
Japan	1986	3 months	Breast milk	21·0	Hojo (1986b)
-			Pasteur. cow's milk	18-9	~ ` ′
			Raw cow's milk	15.0	
			Formula-fed	5.4	
Japan	1991	_	Formula-fed	(3–6)	Tamari et al. (1992)
Japan	1991	_	Colostrum	(10-15)	Yuzo & Mohri (1991)
-			Formula-fed	(3–5)	,
OCEANIA					
New Zealand	1972	1 month	Breast milk	5	Millar & Sheppard (1972)
—			Bottle-fed cow's milk	2	

highest values were obtained in winter and the lowest in summer (Olson & Palmer, 1984). The Se concentration in colostral milk (0·13–0·21 µg/g dry weight) was determined in cows with or without retained placenta (Bostedt & Schramel, 1981) and in healthy cows (Koller et al., 1984). The effect of mastitis on the Se content of cow's milk was studied (Sarudi et al., 1989; Osama et al., 1992). The concentration in milk increased with the presence of mastitis, and also with increasing severity of mastitis (Sarudi et al., 1989), but serum Se levels remained within the normal range (Osama et al., 1992).

SELENIUM INTAKE FOR LACTATING INFANTS

As can be seen in Table 4, there are great variations of the estimated daily intakes of Se in lactating babies from different geographical areas. These values in general are very small compared to those of adults, which amount to about 56 μ g/day in New Zealand (Watkinson, 1974), to 88·3 μ g/day in Japan (Sakurai & Tsuchiya, 1975) and 197 μ g/day in Canada (Thompson *et al.*, 1975). The daily Se intakes of infants, as calculated from daily breast milk consumption, averaged 2·5, 11·8, 24·3, and 94·6 μ g/day in Keshan-disease, low, medium,

and high Se-areas in China, respectively (Levander, 1987; Yang et al., 1989). As a consequence of decrease in dietary intake, a significant decrease in Se in neonates from Helsinki was observed (Kumpulainen et al., 1983b; Alfthan, 1986). Dietary Se intakes of Finnish (Kumpulainen et al., 1983b) and Belgian (Robberecht et al., 1985) breast-fed infants are considerably lower than the lowest level of 'safe and adequate' intake of $10 \mu g/day$ proposed by the National Research Council. But all infants were healthy, gaining weight and height according to the norms of typical Finnish infants (Vuori & Kuitunen, 1979).

On the other hand, infants fed human milk have a higher Se intake than those fed commercially available formula milk or baby foods (Lombeck et al., 1977; Cross et al., 1978; Smith et al., 1982; Oster et al., 1986; Gropper et al., 1990; Tamari et al., 1991). In the case of baby foods, it is surprising because cereals are widely fed as first solid foods, and cereals are considered important sources of Se. However, Se in cereals is lost (7–78%) after dry heating (Morris & Levander, 1970; Higgs et al., 1972) which could explain the lower Se intakes in infants fed baby foods. Therefore, dietary Se intake (μ g/day) of three-month-old Japanese infants fed on infant formula and various milks decreased in the order: human milk (21-0) > pasteurized cow's milk

(18.9) > raw cow's milk (15.0) > infant formula (5.4). So, a two-month-old German infant on cow's milk infant formula alone receives about $7.8~\mu g$ Se/day in its food, while a breast-fed infant receives about $22.4~\mu g$ Se/day (Lombeck et al., 1978). The Se intake of infants fed with commercially available infant formulas and some vegetables is approximately $2-6~\mu g$ /day (Lombeck et al., 1977). Selenium intake of infants fed on the formula seems to be at the intermediate level between Se deficiency and adequate amounts (Yuzo & Mohri, 1991).

These higher intakes in the breast-fed (BF) infant were reflected in higher serum Se compared to a formulafed (FF) infant. So, at one to five weeks of age, the plasma Se content in BF infants was higher than that in cord blood, and in FF infants it was similar to the level in cord blood. A significant positive correlation (r = 0.42, p < 0.05) was found between the Se intake of infants and their serum Se concentration at three months (Smith et al., 1982). Serum Se concentrations were significantly higher in BF infants, than in FF infants. Hatano et al. (1984, 1985) and Gropper et al. (1990) observed significantly lower concentrations of plasma Se in FF than in BF infants, although the Se intake was almost the same in both groups. Consequently, the Se compounds in breast milk seem to be more biologically available for infant nutrition than those in formulas (Hatano et al., 1985; Kumpulainen et al., 1987; Tamari et al., 1991; Yuzo & Mohri, 1991).

CONCLUDING REMARKS

Mature human milk has a Se content lower than transitory milk, and transitory milk lower than colostrum. This fact could be due to differences in protein content, owing to the fact that most of the fraction of Se in milk is associated with the protein fraction.

It can be confirmed that Se in milk, and therefore Se intake and status of newborns, is largely a function of the Se intake of the mother. However, there must be other influential factors such as the form of bioavailability of Se in the diet, as well as interactions with other nutrients.

Decrease with time of Se concentration in mature human milk has been observed in low-Se areas. Body reserves do not seem to be sufficient to maintain the Se level of breast milk.

Human milk has higher Se levels than milk formulas for infants, perhaps due to losses in the latter by processing. Also, Se compounds in breast milk seem to be more biologically available for infants than for those in formulas.

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